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Growth of photoactive CZTS thin film by using spin coating technique

W. M. N. D. Kulasinghe,* W. T. M. A. P. K. Wanninayake, K. M. D. C. Jayathilaka, and L. B. D. R. P. Wijesundera,

Department of Physics, Faculty of Science, University of Kelaniya, Kelaniya

As a promising absorber material in thin film solar cells, the CZTS ($\text{Cu}_2\text{ZnSnS}_4$) has sparked a growing interest over the recent years in the production of low cost solar cells, due to its favorable direct band gap, high optical absorption coefficients, abundance and the non-toxicity of the base materials. A simple and low cost growth technique that offers favorable optoelectronic properties is a prime requirement to lower the cost of solar cells. In this respect, the CZTS thin films have been prepared using a sol-gel method. The solution was made from metal salts of copper (II) chloride (CuCl_2), zinc (II) chloride (ZnCl_2), tin (IV) chloride (SnCl_4), and thiourea ($\text{CS}(\text{NH}_2)_2$) dissolved in a mixture of ethanol and water, mixed in 7:3 ratio, as the solvent. Thin films were grown using the spin coating technique on titanium (Ti) substrates followed by thermal annealing in a nitrogen (N_2) environment. The concentration of the constituents, annealing temperature and the annealing time were varied to identify the optimum photoactive properties of CZTS. Optimum growth conditions of CZTS were investigated based on photoelectric measurements (V_{oc} , I_{sc}) obtained from the dark and light I-V characterization in a photoelectrochemical cell (PEC) containing 0.1 M sodium acetate (CH_3COONa). The results revealed that the best growth conditions were $[\text{CuCl}_2] = 1.7 \text{ M}$, $[\text{ZnCl}_2] = 1.2 \text{ M}$, $[\text{SnCl}_4] = 1.0 \text{ M}$, $[\text{CH}_4\text{N}_2\text{S}] = 7.2 \text{ M}$, spin coating speed = 2000 rpm, spin coating duration = 10 s, annealing temperature = 350°C and annealing duration = 15 min. The maximum V_{oc} and I_{sc} were recorded as 2.820 mV and 26.46 μA respectively with a Fill Factor of 0.273 by an area of 70 mm^2 . Scanning Electron Microscopy (SEM) revealed that the film was uniformly grown on the substrate and the grain size is of the order of $1 \mu\text{m}$. Energy Dispersive Spectroscopy (EDS) revealed that the stoichiometric ratios of $\text{Cu}/(\text{Zn} + \text{Sn})$ and Zn/Sn were 3.20 and 1.22 respectively. Manipulation of the stoichiometry and the handling of the phases of the formation reaction using the proper growth conditions will be subject for further studies.

E-mail: dananjaya.kulasinghe@gmail.com