

## **E1-05 I-V characteristics of ITO/Cu<sub>2</sub>O/metal junctions**

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Cuprous oxide is an inexpensive and non-toxic semiconductor material having the potential for use in low-cost photovoltaic devices. Electrodeposition is a low-cost method to produce thin Cu<sub>2</sub>O films and electrodeposited Cu<sub>2</sub>O has been reported to be n-type. In order to investigate the possibility of using electrodeposited Cu<sub>2</sub>O in solar cells, Cu<sub>2</sub>O/metal junctions were studied using their I-V characteristics.

Thin films of  $\text{Cu}_2\text{O}$  were electrodeposited on Indium tin oxide (ITO) coated glass substrate using an aqueous solution of cupric acetate. Electrodeposition was carried out under potentiostatic condition of  $-250\text{ mV}$  vs SCE at  $55^\circ\text{C}$  for a period of 1 h. These  $\text{Cu}_2\text{O}$  samples were used to make junctions with Al, Hg, Cu, Ag and Au. The I-V plots of these junctions were obtained in dark and when illuminated.

The I-V characteristics of the  $\text{Cu}_2\text{O}/\text{Hg}$  junction show good rectifying properties. The plots for the junction in dark and under illumination indicate that the electrodeposited  $\text{Cu}_2\text{O}$  behaves as an n-type semiconductor. The I-V profile fits into the standard diode equation with  $I_0 = 2.8 \times 10^{-10}\text{ A}$ , ideality factor  $n=0.48$  and series resistance  $R_s = 40\Omega$ . Also, the junction exhibits a contact potential of about  $150\text{ mV}$ .

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